

Communication to the Editor

DYNEMICIN A, A NOVEL ANTIBIOTIC  
WITH THE ANTHRAQUINONE  
AND 1,5-DIYN-3-ENE SUBUNIT

Sir:

The 1,5-diyne-3-ene-containing antibiotics represented by esperamicin<sup>1,2)</sup> and calicheamicin<sup>3)</sup> are receiving increasing attention because of their extremely potent antitumor activity and unusual structures. A unique mechanism of action involving phenyl diradical formation has been proposed for this family of antibiotics<sup>4)</sup>. These antibiotics show extremely strong inhibition of growth of Gram-positive bacteria, especially the recombination-deficient mutants such as *Bacillus subtilis* M45 strain. During the course of our continuing search for new antitumor antibiotics using *B. subtilis* M45, dynemicin A, a novel violet-colored antibiotic was discovered in the fermentation broth of a new *Micromonospora* strain. The antibiotic exhibits very potent antibacterial activity, especially against Gram-positive bacteria, and prolongs the life span of mice inoculated with P388 leukemia. Structural studies revealed that dynemicin A is a unique hybrid of an anthraquinone and an 1,5-diyne-3-ene system. This communication describes the production, isolation, physico-chemical properties, structure, and biological activities of dynemicin A.

The producing organism was isolated from a soil sample collected in Gujarat State, India and was identified as *Micromonospora chersina* sp. nov. No. M956-1. Antibiotic production was carried out in two 200-liter tank fermenters containing 120 liters each of a production medium (soluble starch 1.5%, glucose 0.5%, beet molasses 1%, fish meal 1% and CaCO<sub>3</sub> 0.5%, pH 7.0 before sterilization) at 28°C with agitation (250 rpm) and aeration (120 liters/minute). The antibiotic activity reached a maximum at 92 hours, as monitored by the paper-disc assay using *B. subtilis* PCI 219 as the test organism.

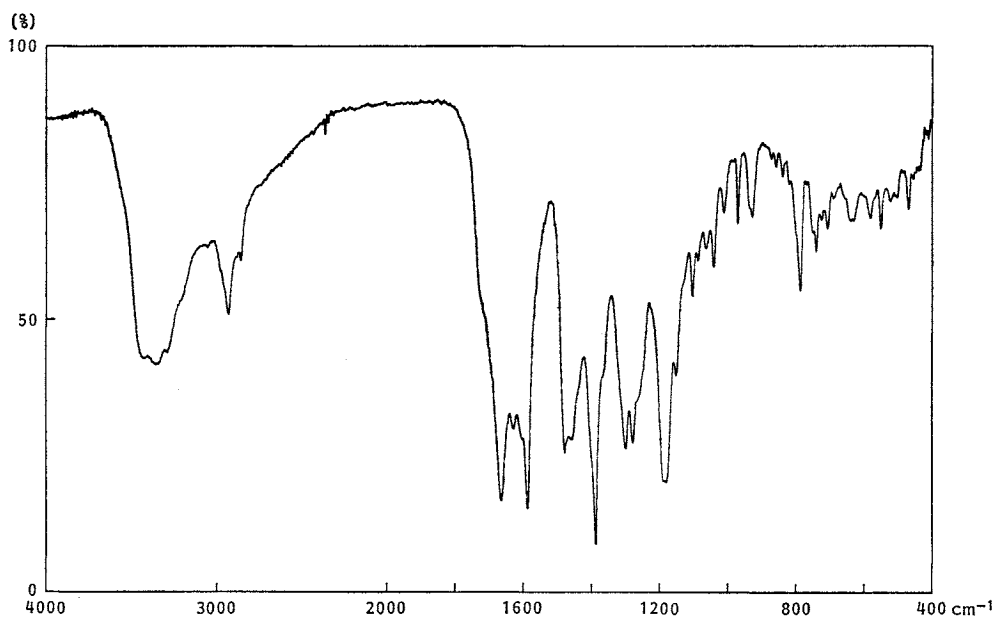
Dynemicin A (1) was isolated from the cultured broth by the following procedure. The whole broth (220 liters) was adjusted to pH 2.0 with 6 N HCl and extracted with BuOH (80 liters). The extract was concentrated *in vacuo* to an aqueous solution (1 liter) which deposited a dark brown precipitate. The solid collected by filtration was dissolved in MeOH (2 liters), combined with the aqueous filtrate and then loaded onto a column of Diaion HP-20 (10 i.d. × 65 cm) previously equilibrated with 70% aqueous MeOH. After being washed with 80% MeOH, the activity was eluted from the column with 80% aqueous acetone. The residue (62 g) obtained upon concentration of the active eluate was rechromatographed on a column of Sephadex LH-20 (4 i.d. × 40 cm) with MeOH elution.

Table 1. Physico-chemical properties of dynemicin A and triacetyldynemicin A.

	Dynemicin A	Triacetyldynemicin A	
Nature	Violet amorphous powder	Orange rods	
MP (°C, dec)	208~210	228~231	
$[\alpha]_D^{25}$	+270° (c 0.01, DMF)	+1,300° (c 0.05, MeOH)	
UV $\lambda_{\max}^{\text{MeOH}}$ nm ( $\epsilon$ )	239 (24,900), 282 (sh), 569 (10,800), 599 (10,100)	244 (40,100), 313 (6,700), 482 (8,100)	
Molecular formula	C <sub>30</sub> H <sub>19</sub> NO <sub>9</sub>	C <sub>36</sub> H <sub>25</sub> NO <sub>12</sub>	
Microanalysis		Calcd for	Found:
		C <sub>36</sub> H <sub>25</sub> NO <sub>12</sub> ·H <sub>2</sub> O:	
		C 63.43,	C 63.20,
		H 3.99,	H 3.75,
		N 2.06	N 2.16
SI-MS $m/z$ , (M+H) <sup>+</sup>	538	664	
TLC <sup>a</sup> (Rf)	0.40	0.33	

SI-MS: Secondary ion mass spectrum. <sup>a</sup> SiO<sub>2</sub>; Xylene - methyl ethyl ketone - MeOH (5 : 5 : 1).

Fig. 1. IR spectrum of dynemicin A (KBr).

Table 2.  $^1\text{H}$  NMR spectrum of triacetyldynemicin A (400 MHz in  $\text{DMSO}-d_6$ ).

Proton No.	Triacetyldynemicin A
4- $\text{CH}_3$	1.25 (3H, d, $J=7.3$ Hz)
11- $\text{OCOCH}_3$ , 15- $\text{OCOCH}_3$ , 18- $\text{OCOCH}_3$	2.33 (3H, s), 2.36 (3H, s), 2.44 (3H, s)
4-H	3.55 (1H, q, $J=7.3$ Hz)
6- $\text{OCH}_3$	3.79 (3H, s)
7-H	4.78 (1H, s)
2-H	5.04 (1H, d, $J=3.8$ Hz)
25-H	6.05 (1H, d, $J=1.3$ Hz)
26-H	6.07 (1H, d, $J=1.3$ Hz)
16-H, 17-H	7.62 (2H, s, $\times 2$ )
10-H	8.03 (1H, s)
1-NH	9.41 (1H, d, $J=3.8$ Hz)
5-COOH	12.37 (1H, br)

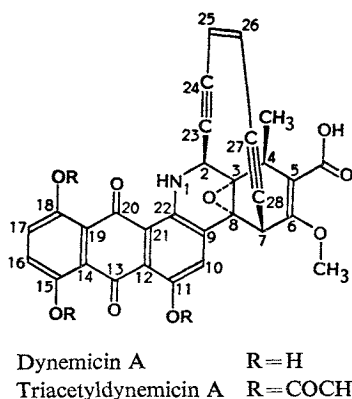
Upon monitoring by TLC (see Table 1), the appropriate eluate was concentrated to yield a dark blue solid (56 mg). This was further purified by preparative TLC ( $\text{SiO}_2$  and the same solvent as above TLC) followed by Sephadex LH-20 chromatography with MeOH elution to yield a homogeneous violet sample of **1** (5.7 mg).

**1** is a violet amorphous solid soluble in DMSO, DMF and dioxane, slightly soluble in  $\text{CHCl}_3$ , EtOAc and MeOH and insoluble in  $\text{H}_2\text{O}$  and *n*-hexane. When treated with acetic anhydride in pyridine, **1** yielded a triacetyl derivative (**2**) with increased solubility. Their physico-chemical properties are summarized in

Table 1. The IR spectrum of **1** (Fig. 1) contains a broad OH/NH absorption band at  $3500\sim 3200\text{ cm}^{-1}$  and carbonyl absorption bands at  $1660$  and  $1630\text{ cm}^{-1}$ . The latter bands suggest a quinone group in the molecule. The IR spectrum of **2** exhibits a strong carbonyl band at  $1770\text{ cm}^{-1}$  in addition to the bands observed for the spectrum of **1**. One methyl ( $\delta$  1.25), three acetyl methyls (2.33, 2.36 and 2.44), one  $\text{OCH}_3$  (3.79), three methines (3.55, 4.78 and 5.04), two olefinic (6.05 and 6.07) and three aromatic protons (7.62 $\times 2$  and 8.03) were observed in the  $^1\text{H}$  NMR spectrum of **2** (Table 2). Corresponding carbon signals were found in the

$^{13}\text{C}$  NMR. The  $^1\text{H}$  NMR spectrum and the UV spectrum of **1** resemble those of  $\epsilon$ -isorhodomyacinone<sup>5,6</sup> suggesting a 1,4,6,9-tetrahydro-anthraquinone or a related chromophore in the molecule. The remaining part of the molecule should have one  $\text{CH}_3$ , one  $\text{OCH}_3$ , three  $>\text{CH}$ , two  $-\text{CH}=\text{}$ , two  $>\text{C}=\text{}$ , six quaternary carbons and one carboxyl carbon. Among the quaternary carbons, four carbons appeared at  $\delta$  88.8, 89.6, 97.3 and 99.4 strongly suggesting a conjugated diene system from spectral comparison with esperamicin. The complete structure was elucidated by X-ray crystallography of crystalline **2'** (Fig. 2).

Fig. 2. The structures of dynemicin A and triacetyldynemicin A.



Dynemicin A (**1**) and its triacetate (**2**) showed extremely strong activity against Gram-positive bacteria as shown in Table 3. Gram-negative bacteria, anaerobic bacteria and fungi are considerably less sensitive to both compounds. On the whole, **2** is two to eight times more potent than **1** against the organisms tested. **1** exhibited significant *in vivo* activity against *Staphylococcus aureus* Smith infection in mice with the  $\text{PD}_{50}$  being 0.13 mg/kg by ip administration. No toxic signs were observed in the mice after administration of 5 mg/kg (ip) of **1**. Both compounds showed marked cytotoxic activity against B16 melanoma, Moser human carcinoma, HCT-116 human carcinoma and the normal and vincristin-resistant P388 leukemia cells with  $\text{IC}_{50}$  of 0.004~0.005  $\mu\text{g}/\text{ml}$ . In *in vivo* tests, **1** and **2** produced significant prolongation of life span of mice inoculated with P388 leukemia and B16 melanoma (Table 4). The active dose ranges were rather broad though the T/C values were maintained at not very high levels.

It is apparent that dynemicin A is a new member of the esperamicin/calicheamicin family of antibiotics in terms of possessing the unique 1,5-dien-3-ene unit. However, it is distinctly different from the preceding antibiotics in possessing the violet-colored chromophore of a substituted anthraquinone. It should also be noted that unlike the esperamicin antibiotics, **1**

Table 3. Antibacterial spectra of dynemicin A and triacetyldynemicin A.

Organism	MIC ( $\mu\text{g}/\text{ml}$ )	
	Dynemicin A	Triacetyldynemicin A
<i>Staphylococcus aureus</i> FDA 209P	0.000013	0.0000063
<i>S. aureus</i> Smith	0.000025	0.0000063
<i>S. epidermidis</i> D153	0.0000063	0.0000031
<i>Micrococcus luteus</i> PCI 1001	0.0008	0.0002
<i>Bacillus subtilis</i> PCI 219	0.0000063	0.0000031
<i>Escherichia coli</i> NIHJ	0.05	0.0063
<i>Klebsiella pneumoniae</i> D11	0.0063	0.0008
<i>Pseudomonas aeruginosa</i> A9930	0.025	0.0063
<i>Proteus vulgaris</i> A9436	0.0063	0.0031
<i>Clostridium difficile</i> A21675	0.0063	0.0031
<i>Bacteroides fragilis</i> A22693	0.2	0.1
<i>Candida albicans</i> IAM 4888	12.5	0.4
<i>Cryptococcus neoformans</i> D49	12.5	0.8
<i>Aspergillus fumigatus</i> IAM 2530	6.3	0.1
<i>Trichophyton mentagrophytes</i> D155	12.5	0.4

<sup>†</sup> Full details of the structure determination will be forthcoming; M. KONISHI, H. OHKUMA, T. OKI, H. KAWAGUCHI and J. CLARDY.

Table 4. Antitumor activity of dynemicin A.

	P388 leukemia		B16 melanoma	
	Dose qd 1→3, ip (mg/kg/day)	T/C (%)	Dose qd 1→9, ip (mg/kg/day)	T/C (%)
Dynemicin A	1.0	135 <sup>a</sup>	1.0	159 <sup>a</sup>
	0.5	130	0.5	137
	0.25	135	0.25	137
	0.13	130	0.13	122
	0.063	130	0.063	141
	0.031	130	0.031	122
Mitomycin C	3.0	200	2.0	222
	1.0	135	1.0	152
	0.3	140	0.5	133
	0.1	115	0.25	111

\* T/C  $\geq$  125 means significant antitumor effect.

exhibited significant *in vivo* antibacterial activity and low toxicity. The anthraquinone chromophore moiety of dynemicin A is presumed to play an important role in its biological activity and further studies on the subject will be pursued.

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#### References

- 1) KONISHI, M.; H. OHKUMA, K. SAITOH, H. KAWAGUCHI, J. GOLIK, G. DUBAY, G. GROENEWOLD, B. KRISHNAN & T. W. DOYLE: Esperamicins, a novel class of potent antitumor antibiotics. I. Physico-chemical data and partial structure. *J. Antibiotics* 38: 1605~1609, 1985
- 2) GOLIK, J.; G. DUBAY, G. GROENEWOLD, H. KAWAGUCHI, M. KONISHI, B. KRISHNAN, H. OHKUMA, K. SAITOH & T. W. DOYLE: Esperamicins, a novel class of potent antitumor antibiotics. 3. Structures of esperamicins A<sub>1</sub>, A<sub>2</sub>, and A<sub>1b</sub>. *J. Am. Chem. Soc.* 109: 3462~3464, 1987
- 3) LEE, M. D.; T. S. DUNNE, C. C. CHANG, G. A. ELLESTAD, M. M. SIEGEL, G. O. MORTON, W. J. MCGAHREN & D. B. BORDERS: Calicheimins, a novel family of antitumor antibiotics. 2. Chemistry and structure of calicheimicin  $\gamma_1^1$ . *J. Am. Chem. Soc.* 109: 3466~3468, 1987
- 4) ZEIN, N.; A. M. SINHA, W. J. MCGAHREN & G. A. ELLESTAD: Calicheimicin  $\gamma_1^1$ : An antitumor antibiotic that cleaves double-stranded DNA site specifically. *Science* 240: 1198~1201, 1988
- 5) OKI, T.; A. YOSHIMOTO, Y. MATSUZAWA, T. TAKEUCHI & H. UMEZAWA: Biosynthesis of anthracycline antibiotics by *Streptomyces galilaeus*. I. Glycosidation of various anthracyclines by an aclacinomycin-negative mutant and biosynthesis of aclacinomycins from aklavinone. *J. Antibiotics* 33: 1331~1340, 1980
- 6) MATSUZAWA, Y.; A. YOSHIMOTO, T. OKI, H. NAGANAWA, T. TAKEUCHI & H. UMEZAWA: Biosynthesis of anthracycline antibiotics by *Streptomyces galilaeus*. II. Structure of new anthracycline antibiotics obtained by microbial glycosidation and biological activity. *J. Antibiotics* 33: 1341~1347, 1980